# Use of ITC/DSC for the studies of biomolecular stability and interaction of excipients with proteins

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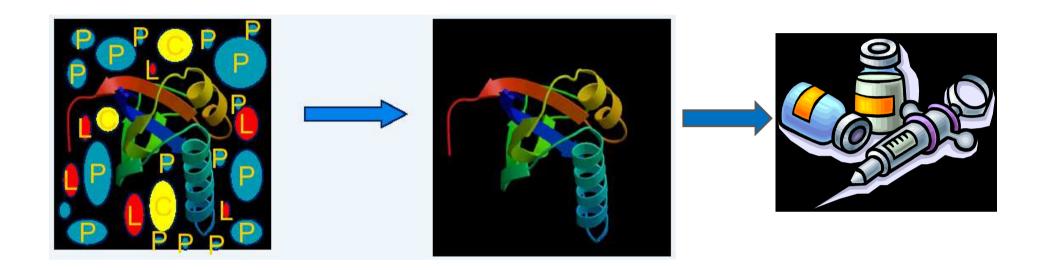


#### Outline

- Brief intro to the specifics of protein stability profiling and optimization
- Key benifits of the application of lable-free biophysical technologies
- Introduction into microcalorimetric technology principles and the kind of information they can provide
- Examples on the use of DSC and ITC



### Complex task of protein stabilization



Chemical stress (pH, ionic strength, etc.)

Physical stress (UF/DF, temperature)

Freeze-thaw-induced stress



### Numerous choices of excipeints available for empirical screening

#### **Excipient types**

- ✓ Buffering agents
- ✓ Amino acids
- ✓ Osmolytes
- ✓ Sugars and carbohydrates
- ✓ Proteins and polymers
- ✓ Salts
- ✓ Surfactants
- ✓ Chelators and antioxidants
- ✓ Preservatives
- √ Specific ligands

#### MOA:

- changes in bulk solutions properties
- specific interaction with protein
- inhibition of a degradation pathway



#### Empirical vs rational approaches

Proteins exhibit a wide range of degradation phenomena.

Protein stability depends on many parameters.

Profiling and optimization of protein stability remains a combination of trail-and-error and rational approaches.



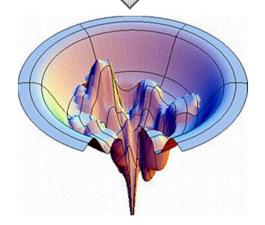


Profiling and optimization of protein stability



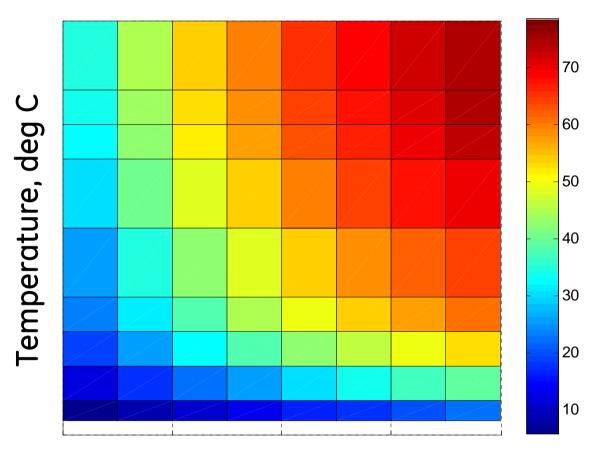
Data analysis for trends and sweet spots

Optimization of protein stability





### Looking for trends and sweet spots: Empirical phase diagram



Understanding of factors (intrinsic and extrisic) critical to a protein stability on the molecular level is needed for implementaion of the rational approach to optimization of protein stability





### Need for orthogonal approach: No single technique works well in all cases

Every single technique has limitations associated with the underlying measuring principles. The results can be model and method dependent.

Convergence of evidence from several techniques is often needed to gain insights into conditions critical to protein stability and to deduce the mechanism of protein-excipient interaction



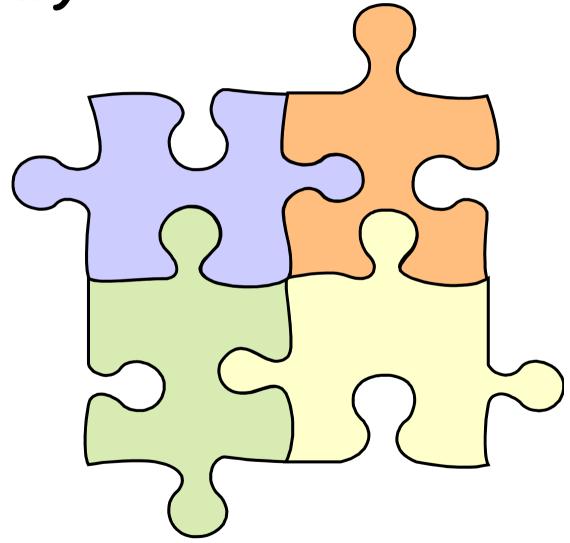
### Biophysical techniques

#### **Upsides of biophysical techniques:**

- first principle data produced with fewer assumptions and simplifications
- possibility to cross-correlate experimental findings to assure their reliability
- better control over experimental parameters and the reacting species better possibility to eliminate artifacts



### Microcalorimetry





### Microcalorimetry

**ITC**, Isothermal Titration Calorimetry

- Heat of interaction is measured
- Binding reaction of two components
- One temperature

**DSC**, Differential Scanning Calorimetry

- Thermal denaturation of protein is monitored
- Thermal stability of proteins at different conditions (buffers, excipients, adjuvants)
- Temperature is ramped

### MicroCal™ systems

Differential scanning calorimetry (DSC) delivers information on protein stability

Isothermal titration calorimetry (ITC) allows to assess binding affinity and stoichiometry in one simple assay



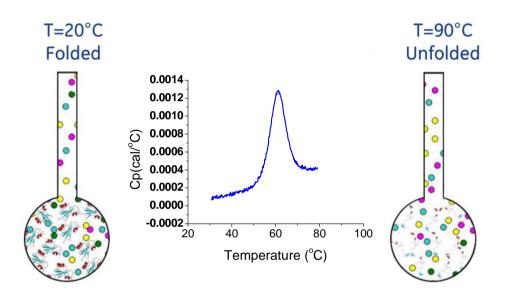






#### What does DSC actually measure?

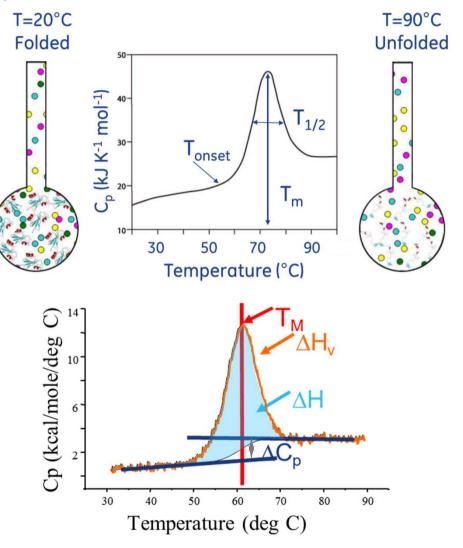
- Measures heat capacity change associated with thermal unfolding
- Provides a thermogram which is a qualitative and quantitative fingerprint of protein unfolding profile





### DSC thermogram provides multiple descriptors of protein thermal stability

- T<sub>m</sub> is the thermal transition midpoint
- $\Delta H$ ,  $\Delta H_{vH}$ ,  $\Delta C_p$
- T<sub>onset</sub> unfolding start
- $\Delta T_{1/2}$  homogenity of population
- Domain resolution
- Assessment of oligomerization and aggregation propensity
- Percent reversibility of the transition
- Kinetics of irreversible process





### Advantages

- Generic: almost all transitions have an enthalpy change
- Domain resolution
- Broad dynamic range of attainable temperatures, scan rates (10÷240 °C/h) and possibility for re-scans
- Temperature resolution to 0.2 °C
- No optical probes
- No labelling, allows to study sample as it is
- Can run on turbid solutions



### DSC applications

#### DSC can be used to assess:

- Thermodynamic and kinetic stability
- Shelf life/stability
- Contribution of different groups to the stability
- Effect of buffer and storage condition
- Protein-ligand interaction



### MicroCal VP-Capillary DSC System

Automated system medium throughput screening

Up to 50 samples/ day; down to ~50 μg of a protein per scan

96-well plate format



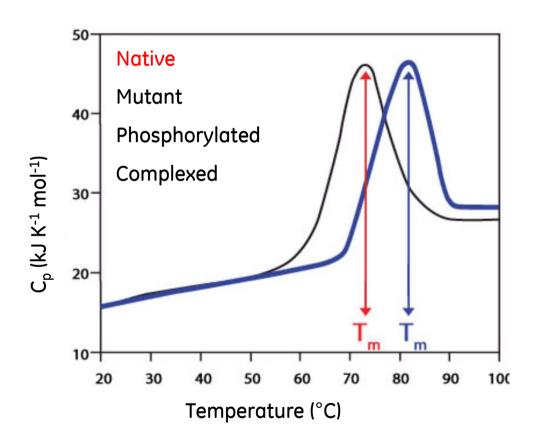
- ✓ Measures the temperature (Tm) associated with a thermal unfolding of a protein
- √Tm is an indicator of stability



#### **Concentration Requirements**

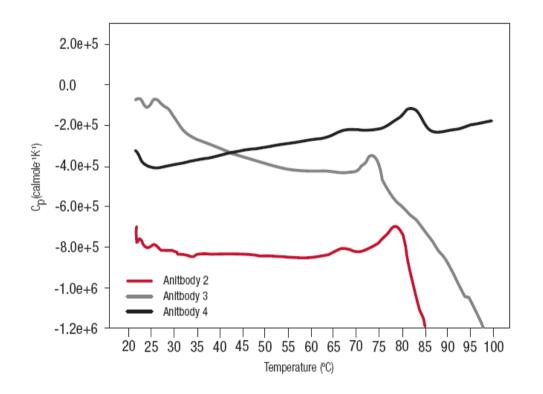
- ✓ Minimum concentration 0.02 mg/ml (~10 µg)
- ✓ As starting point min 0.1-0.2 mg/ml
- ✓ Maximum concentration 5 10 mg/ml

### Comparison of native, altered and mutant forms





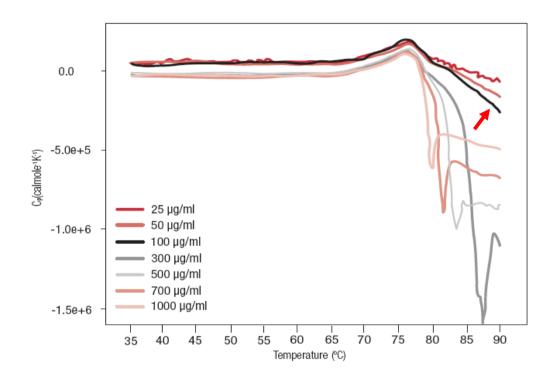
#### Construct selection: IgG1 Thermal Stability



### Antibody 3 would be removed from the drug development pipeline

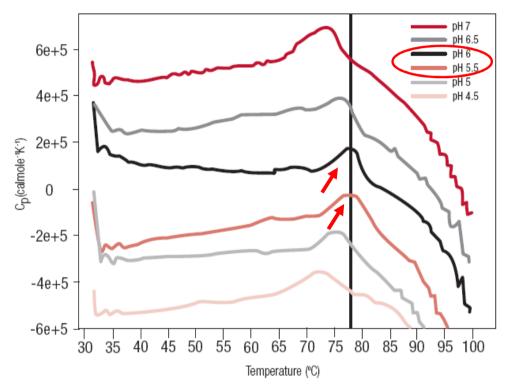


### Colloidal stability at different concentrations: IgG1 Thermal Stability





### Screening different pHs: IgG1 Thermal Stability



Best pH 6 or 6.5



#### How Does DSC Compare?

Results from a pH primary screen of a therapeutic antibody

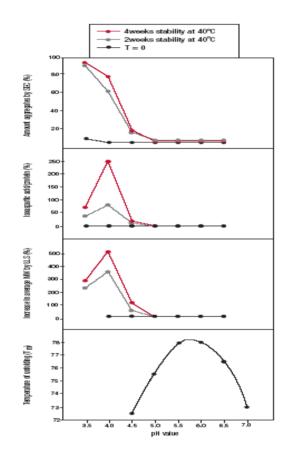
DSC was the most accurate and fastest predictor of suitable formulations

Size Exclusion Chromatography

Isoaspartate Formation CE

Laser Light Scattering

DSC  $T_M$ 

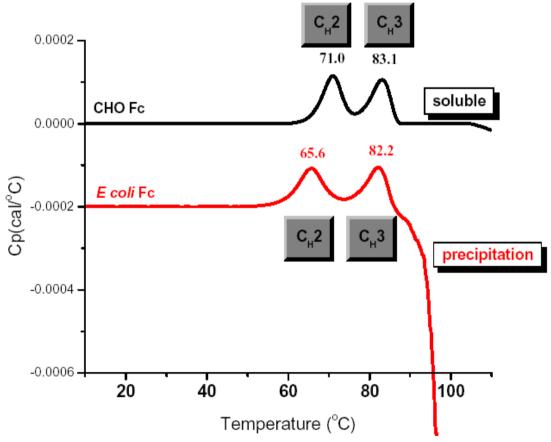


Red=4 weeks at 40 °C Grey=2 weeks at 40 °C Black=T=0



### Sugar effect on stabilization of Fc domains

Two different protein expression systems



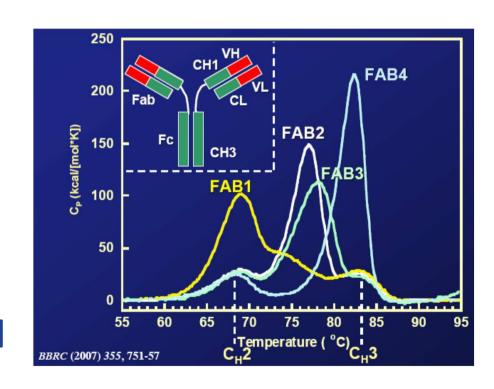


### Stability contributions of individual domains identified

Individual domains are often masked

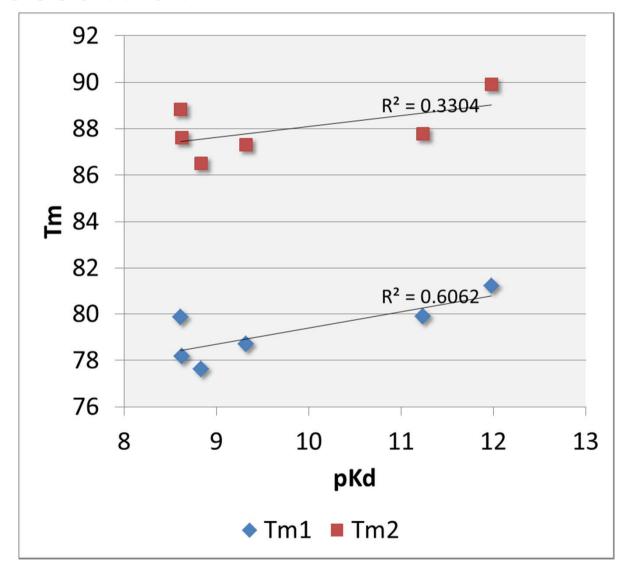
DSC uniquely identified individual stability contribution

Increased stability correlates with increased functional protein fraction



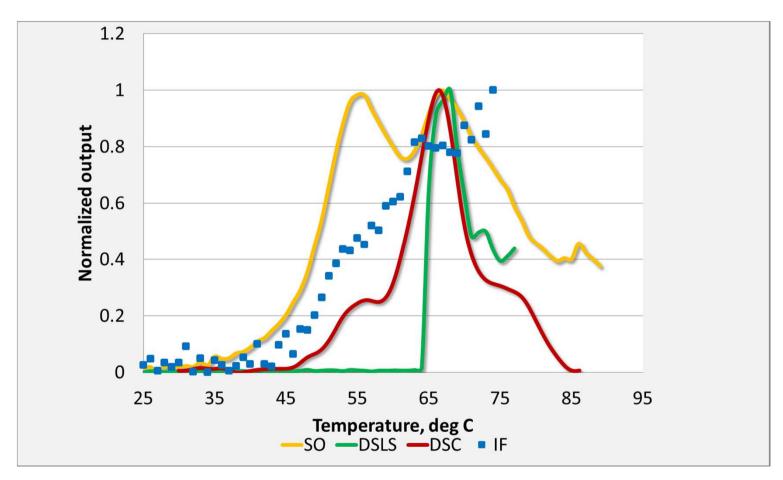
### mAb binding affinity to anti-Fc Ab correlates best with Tm1

Binding affinity determined with Biacore™ T200.



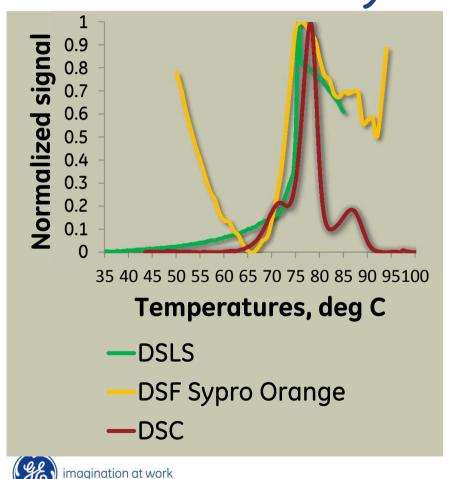


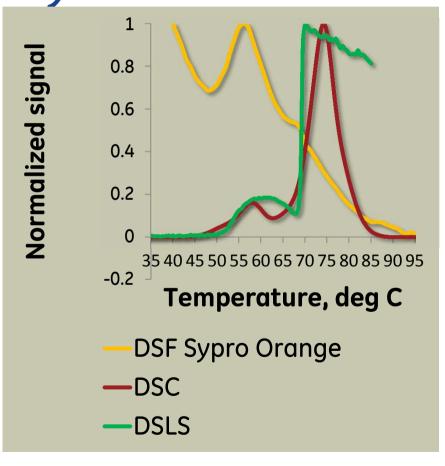
### Domain resolution with different thermal stability techniques





Robustness of baseline definition and domain resolution differs between thermal stability assays



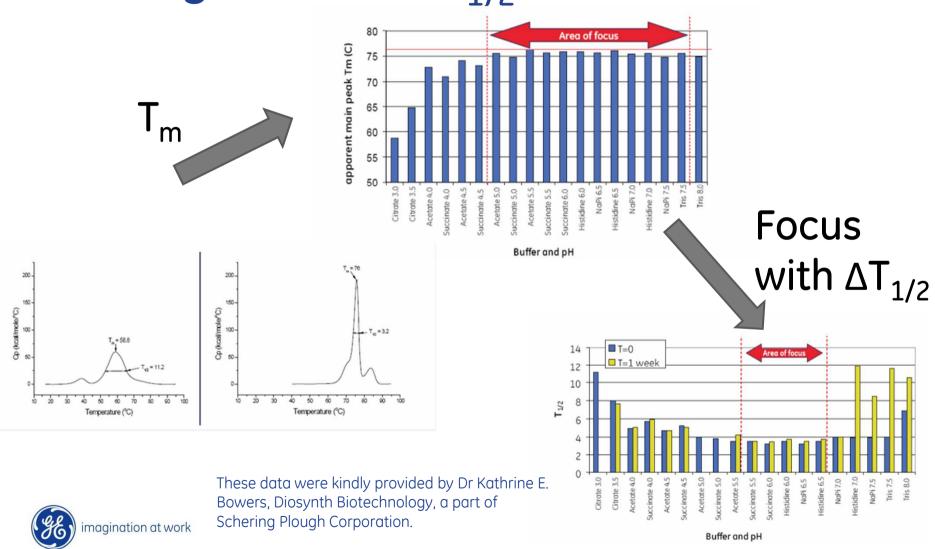


### Addressing complexity with DSC

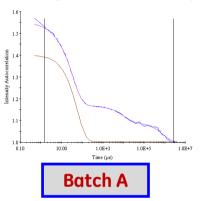
Individual domain stabilization
Using multiple descriptors
Using thermogram as a fingerprint

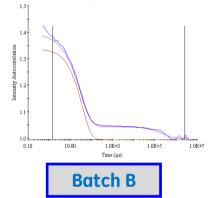


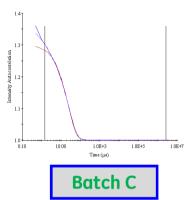
### Pre-formulation decisions funnel utilizing Tm and $\Delta T_{1/2}$

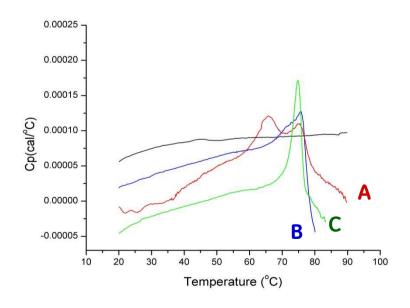


### Thermogram shape: indication of heterogeneity of higher order structure









#### **DLS and DSC**:

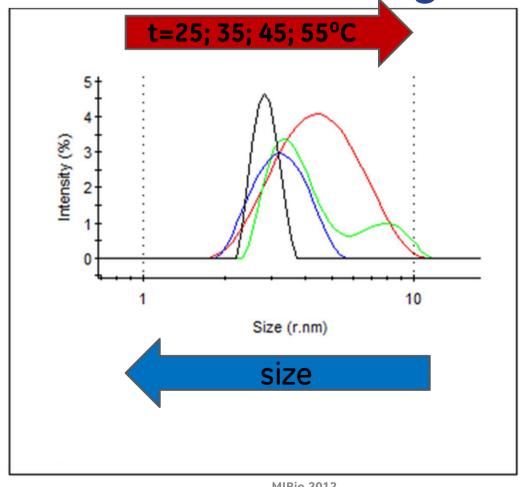
- ➤ Batch A: conformational heterogeniety, thermal lability.
- Batch B: decreased heterogeneity.
- Batch C: homogenous and most stable.

## Thermogram shape: indication of heterogeneity. On-going dissiciation of protein oligomers?

—Protein X in PBS pH 7.4 —Protein X in NaAcO pH 5 8 Cp, kcal/mol/deg C 50 60 70 80 40 Temperature, deg C

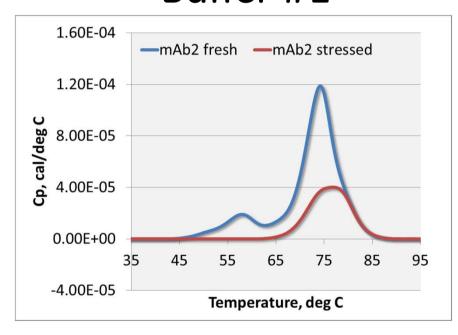


Convergence of DLS and DSC data: in PBS Protein X undergoes temperature induced dissociation of oligomers

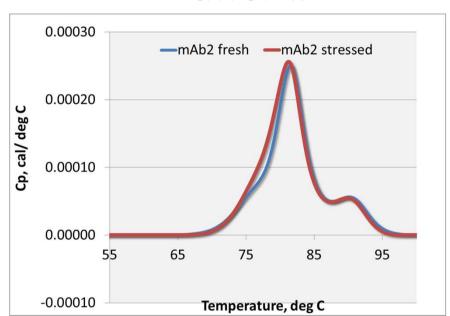


### Thermogram shape and area as means to assess protein viability

Buffer #1



#### Buffer #2

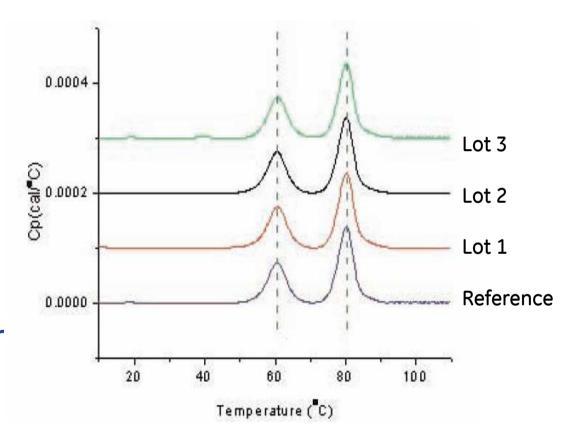




### Easily Assess Biocomparability with DSC

Three lots manufactured at different sites

DSC verifies no difference in stability and solubility betweer lots



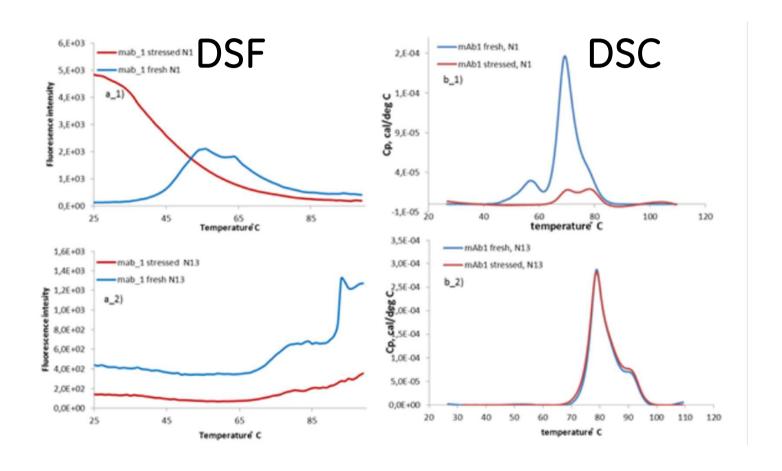


### Advantages

- Generic . Almost all transitions have an enthalpy change
- Broadest dynamic range of attainable temperatures, scan rates and possibility for re-scans
- Temperature resolution to 0.2 °C
- No optical probes
- No labelling, allows to study the sample as it is
- Possible to measure on turbid solutions



### Robustness of qualitative and quantitative comparison of fresh and stressed mAb1 samples



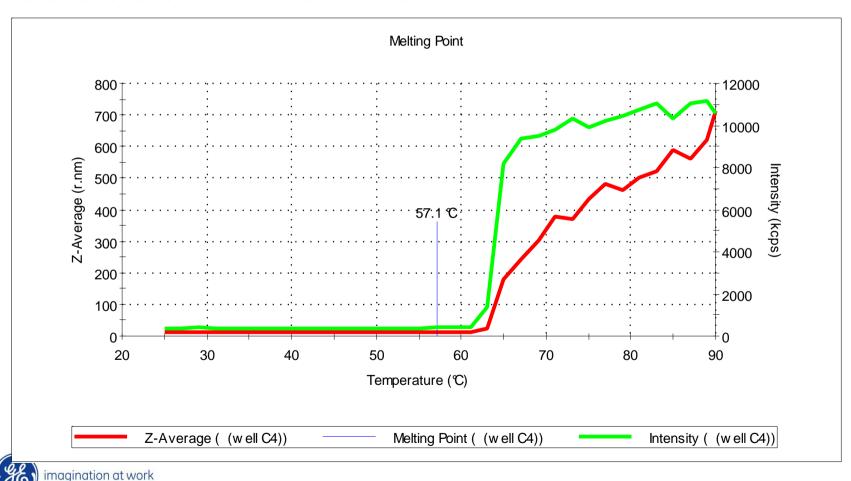


## Screening with DSC – testing the effects of diverse excipients with no interference in instrumental readout

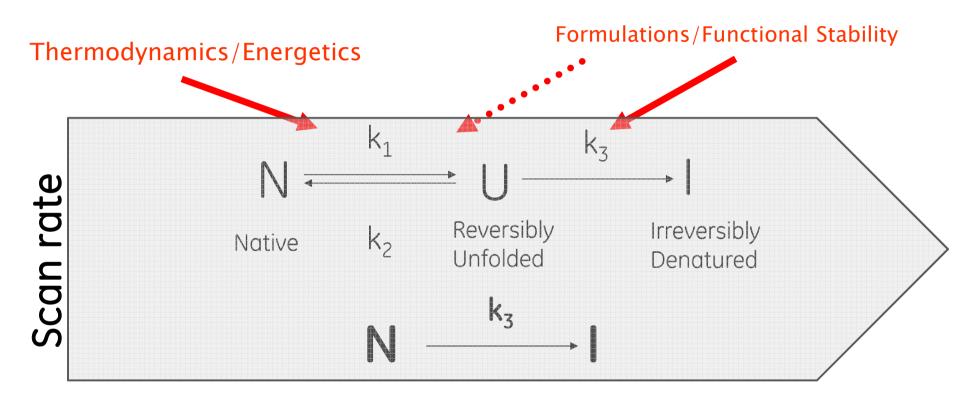
DSC can screen a wide variety of excipients

	Excipient	Tm (°C)	
Control	-	48.1	
Sugars	Manitol	46.7	
	Glucose	49.6	
Polymers /	PEG (300)	49.4	
Polyols	Ethanol	48.7	
	Ethanol	43.8	
Salts	NaCl 53		
	CaCl <sub>2</sub>	41.1	
Surfactants	Pluronic F68		
	Tween 80	45.8	
Effects of Excipients on Stability of IL-R			

## Some excipients can corrupt thermal stability assays based on optical readout: Tween 20 in DLS

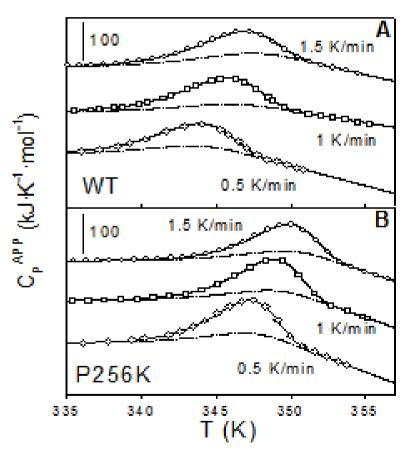


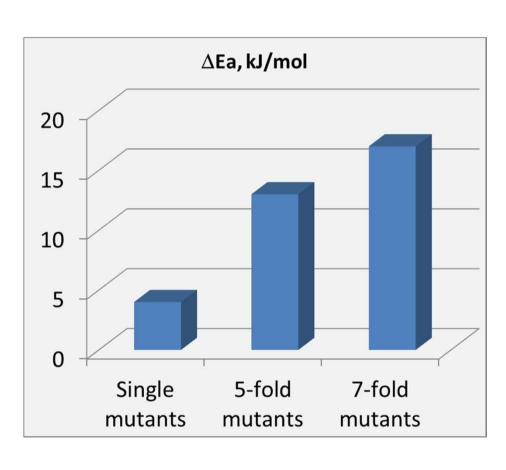
## Protein denaturation in non-simplified form



Irreversible unfolding N->D, where  $k_3$  is a first-order kinetic constant

## Kinetic stabilization confirmed through scan rate dependence of Tm in DSC

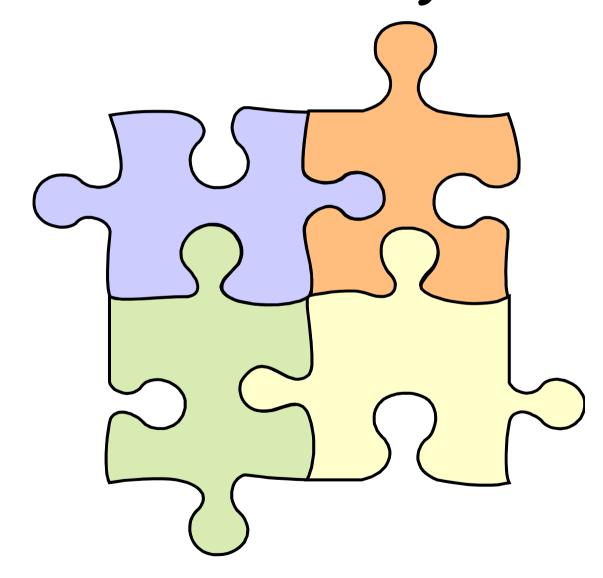






Rodriguez-Larrea et al. J. Mol. Biol. (2006) 360, 715-724

### **Isothermal Titration Calorimetry**





## Isothermal titration calorimetry (ITC) provides

Assays for biological activity by monitoring in one simple assay:

- Binding affinity
- Stoichiometry of binding

Assays for formulation development by monitoring:

• Protein – excipient interactions



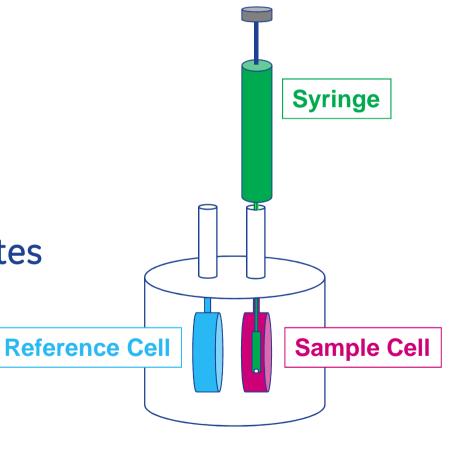
MicroCal™ iTC200

#### How does ITC work?

Measures heat of interaction

#### Single ITC experiment

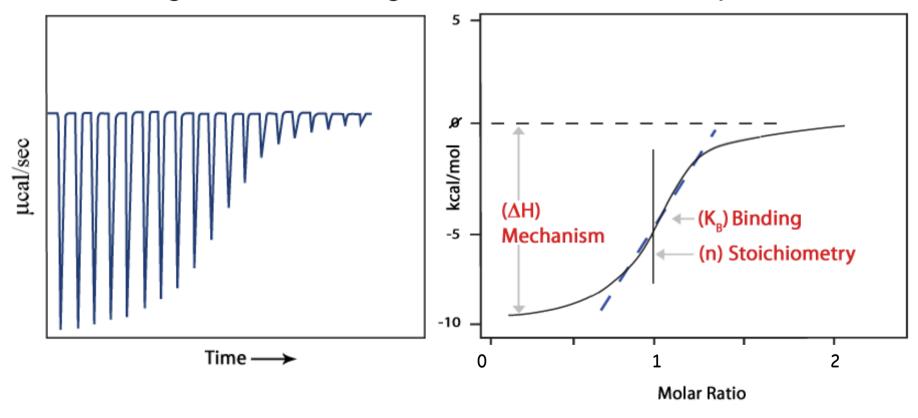
- Affinity
- Binding mechanism
- Number of binding sites



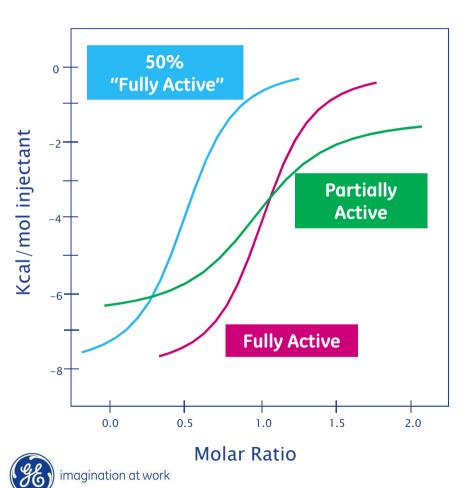


### Fitting ITC data

For a binding isotherm, integrate the area for each peak



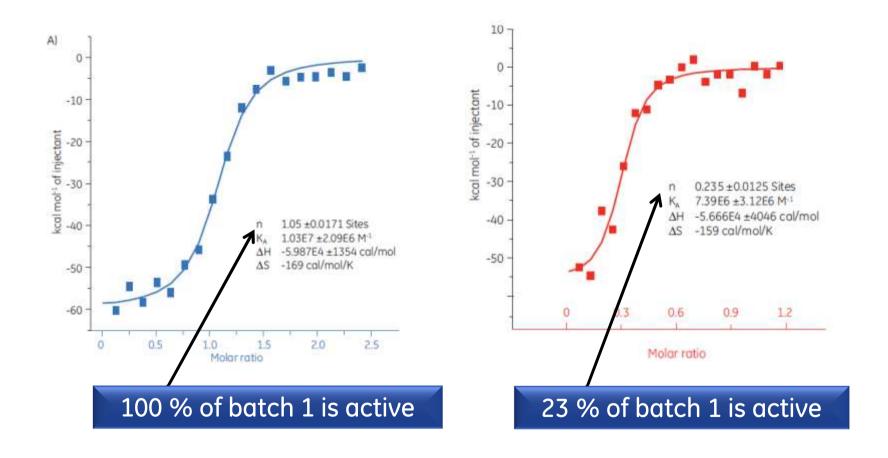
### Affinity and stoichiometry



#### **Protein Quality**

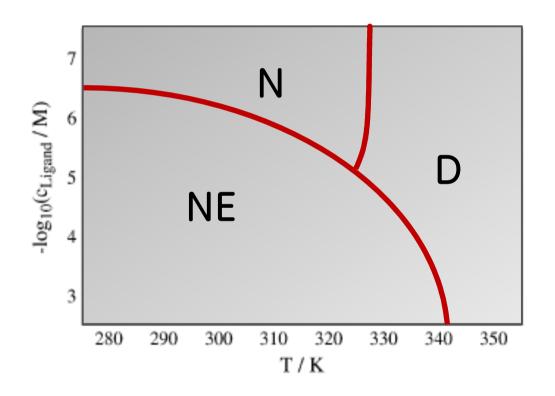
- Measure active concentrations
- Compare protein batches
- Investigate MOA

### Protein quality/assay development





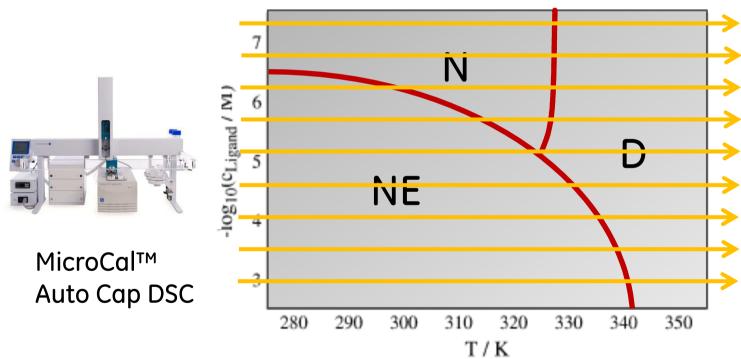
# Getting a bigger picture on protein thermal stability and interaction with excipient: Phase diagram





## Mapping stability profile of a protein







Profiling protein-ligand

interaction

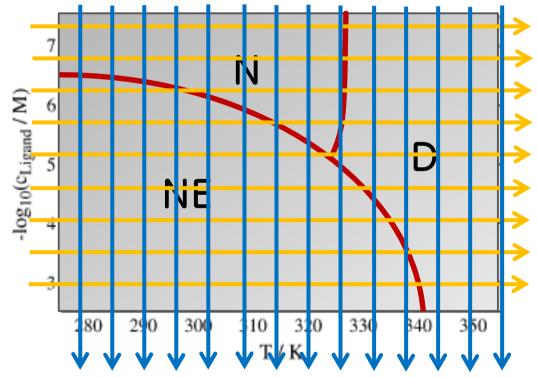


MicroCal™ Auto iTC200



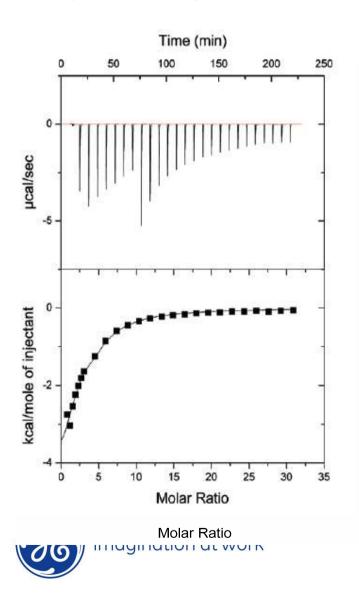


MicroCal™ Auto Cap DSC





### Explore protein - excipient binding



Binding of polysorbate-80 to Protein X

Binding saturation of ~10 moles of polysorbate-80 per mole of Protein X

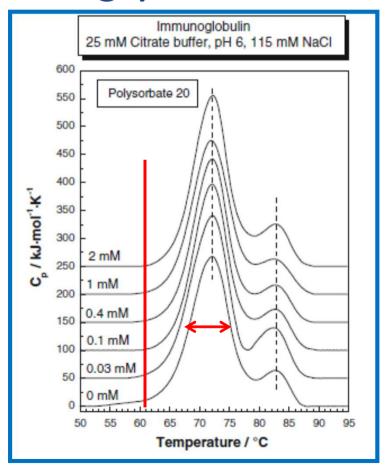
Weak interaction: polysorbate-80/Protein X complex more likely to dissociate in vivo

ITC data suggests minimum excipient concentration needed to stabilize Protein X in formulation

MicroCal Appl Note 28-9613-26 AA

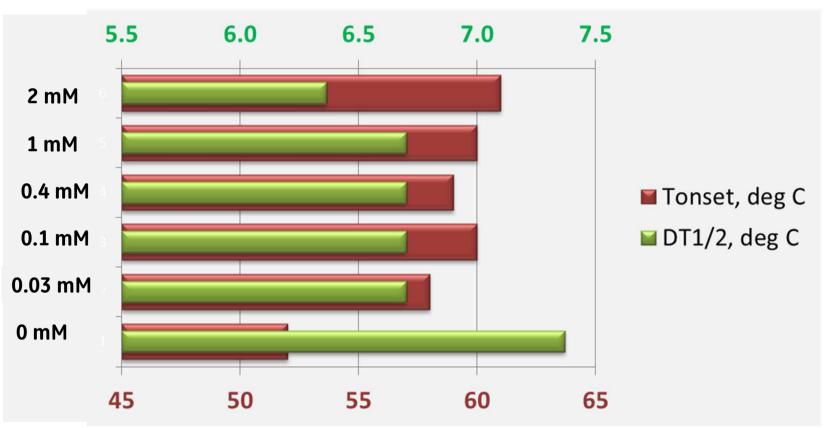
MIBio2012

# Quality and quantity: DSC thermograms offer multiple descriptors of Ab1 unfolding profile





## T<sub>m</sub>s unaffected **BUT** Tween 20 affects cooperativity of Ab1 unfolding





## Screening - evaluation of preservatives for IL-1R

DSC quickly and easily identifies most stable conditions for optimal liquid formulations

DSC stability indications correlate well with SEC results

	<u>T<sub>m</sub></u> <u>% Aggregation by SE</u>	
Control	50.8	1.5
Phenol	50.3	3.0
m-Creosol	48.4	5.1
Benzyl alcohol	45.2	16.5

## ITC and DSC used to study effect of preseravtives on an Ab

MicroCal Appl Note 28-9613-26 AA



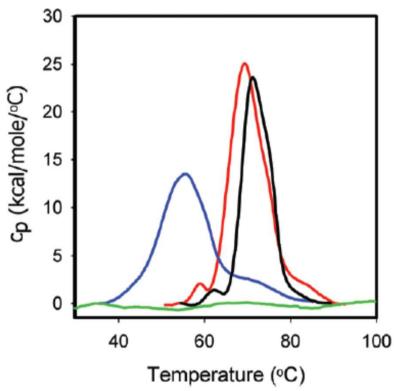
### Effect of phenol on Protein X formulations

✓ Phenol was found to have a decreased antimicrobial activity in the presence of Protein X

MicroCal Appl Note 28-9613-26 AA

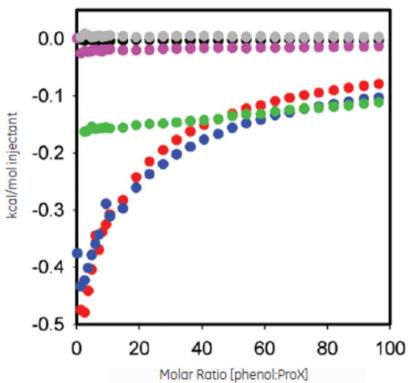


## Phenol binds to Protein X and stabilizes it against thermal unfolding



DSC curves: pH 5.7 (free protein - red line; protein+phenol black line), pH 4.5 (blue), pH 3.5 (green).





ITC data on protein X titration with phenol: pH 5.7 (red line), pH 4.5 (blue), pH 3.5 (green). Rest of the data represents reference titrations.

MicroCal Appl Note 28-9613-26 AA

#### Conclusions

Profiling and optimization of protein stability is a complex task which needs to be approached by several orthogonal techniques.

Application of ITC and DSC can help in most steps of Biopharmaceutical development from initial selection and optimization of the leads, to formulation and QC.

As first principle biophysical techniques ITC and DSC can deliver insightful details on protein stability profile and protein-ligand interactions.

DSC is well suited for smaller scale buffer/excipient screens, DSF assay validations and in-depth characterization of protein and protein-ligand interactions.



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