

# The Secret Life of a Pickering Emulsion

Paul S. Clegg



**SOFI** **CDT**

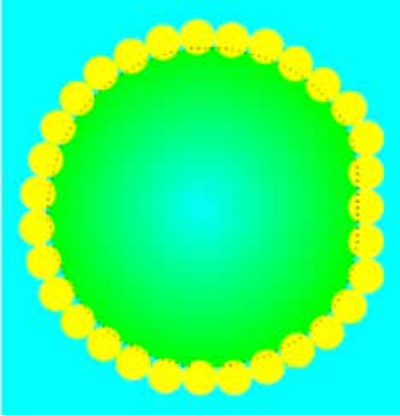
EPSRC Centre for Doctoral Training in  
Soft Matter and Functional Interfaces

**syngenta**

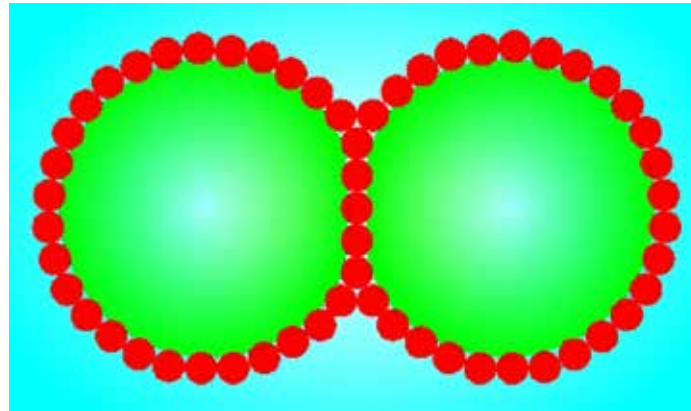
**ECFP** 

# Outline

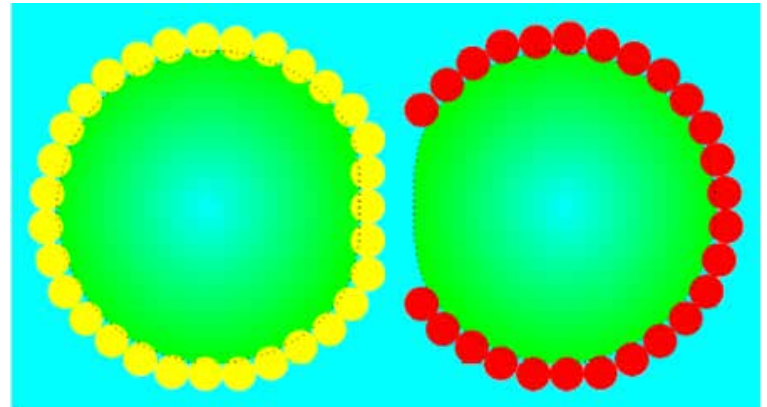
1. Pickering emulsions



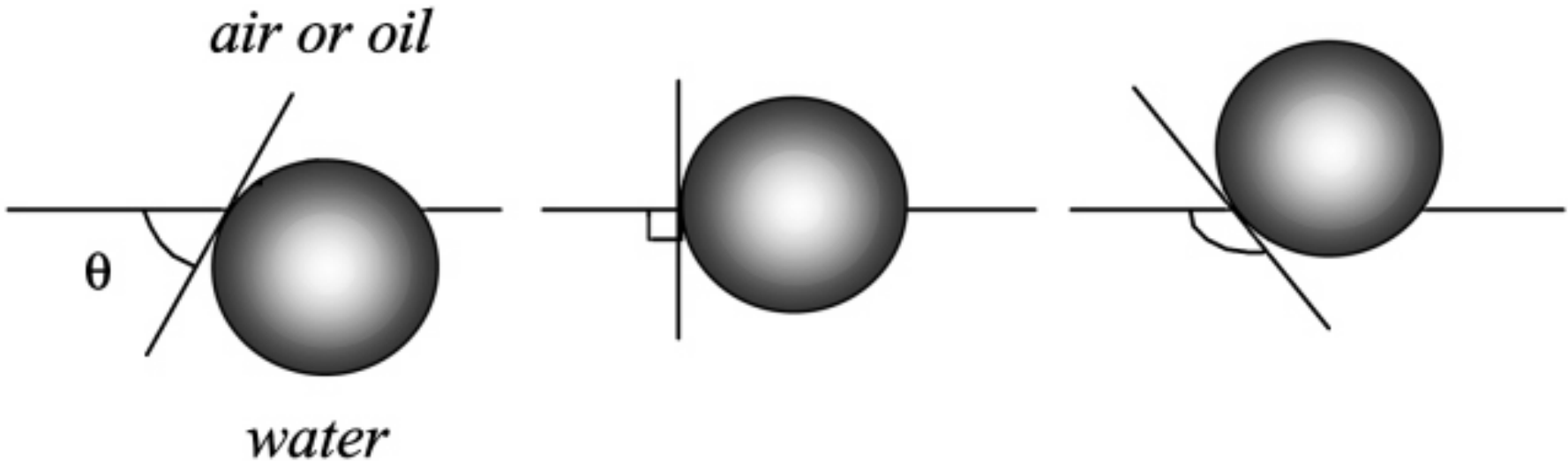
2. Making & breaking bridges



3. The secret life of  
Pickering emulsions

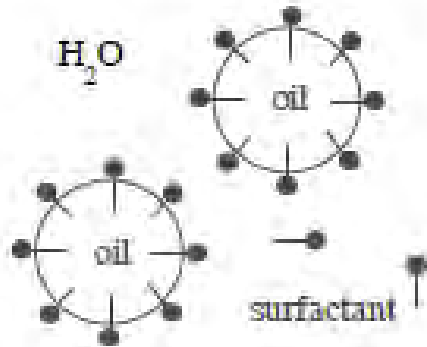


# Particles at liquid-liquid interfaces



# Emulsions

Conventional emulsion



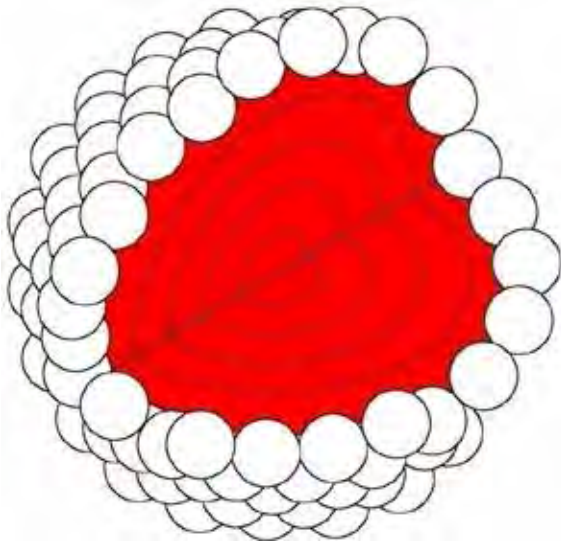
Surface active agent

Trapping energy  
 $\Delta E \sim k_B T$

i.e. surfactants  
hop on and off

“Infinitesimal” increase  
in area is covered by  
fresh surfactant

$\gamma$  is decreased



Pickering emulsion

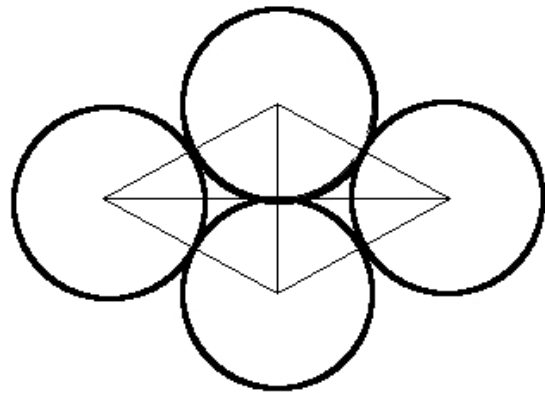
Trapping energy  
 $\Delta E > 10,000 k_B T$

i.e. particles  
irreversibly trapped

“Infinitesimal” increase  
in area cannot be  
covered by mesoscale  
colloids

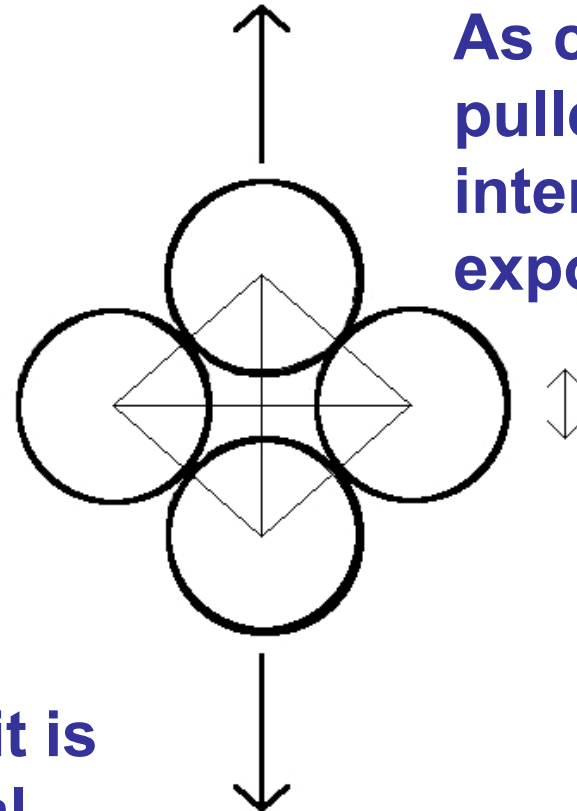
$\gamma$  unchanged or  
inappropriate

# Origin of interfacial elasticity



When an interface of close-packed colloids is strained it is predominantly the interfacial tension that leads to the elastic properties.

Young's Modulus  $E \sim \gamma / d$

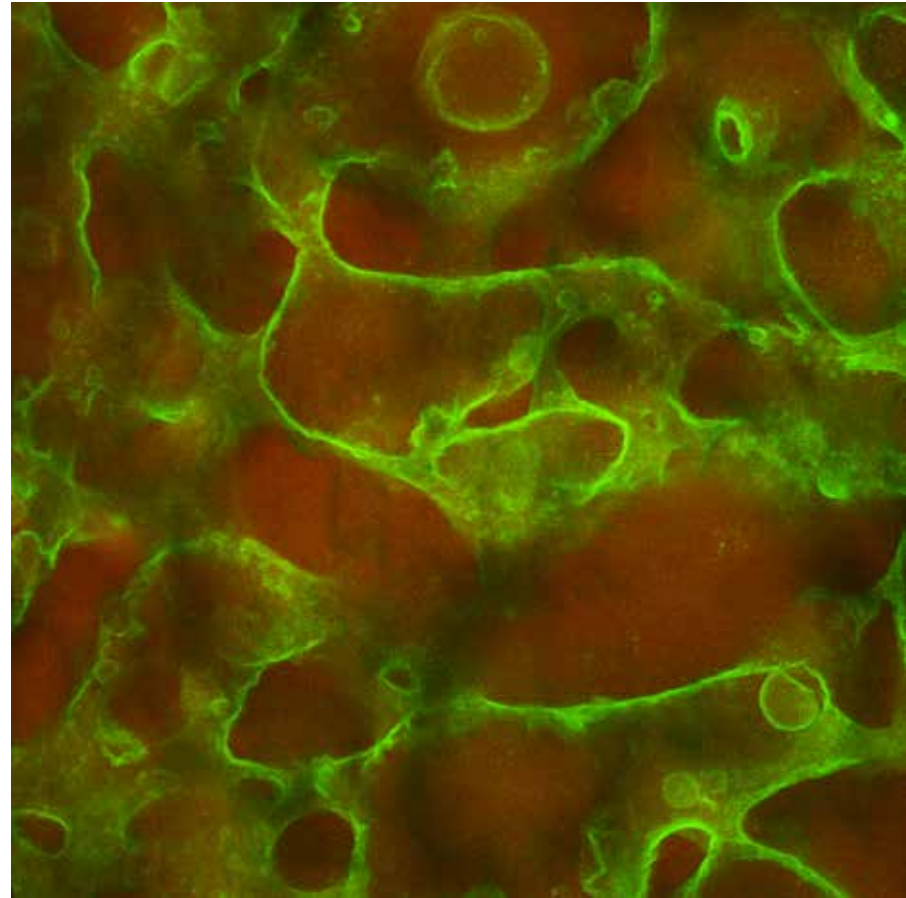
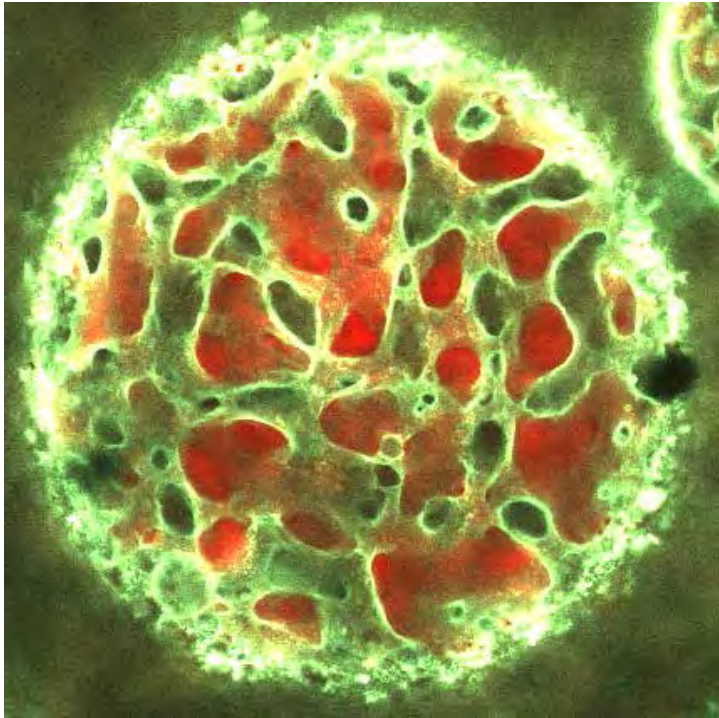


As colloids are pulled apart new interface is exposed

D. Vella *et al*, Europhys. Lett. **68**, 212 (2004)

# Bijels: bicontinuous domains

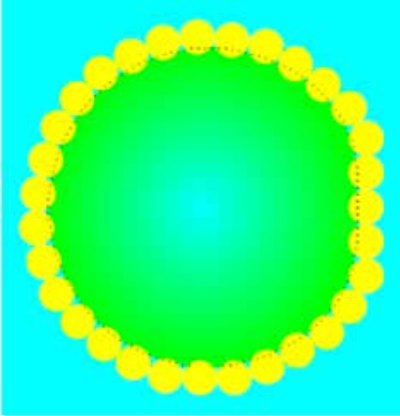
Tavacoli et al Adv. Funct. Mater.  
21, 2020 (2011)



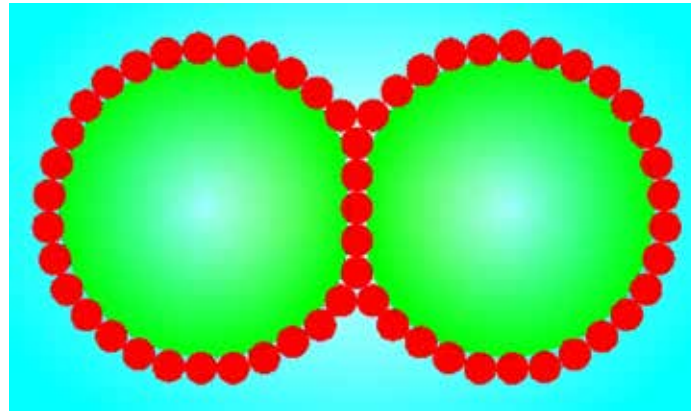
liquids separate via spinodal decomposition  
particles *jam* on interfaces

# Outline

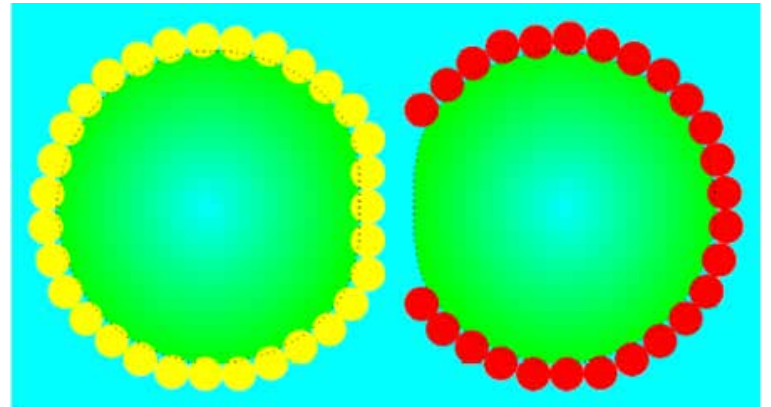
1. Pickering emulsions



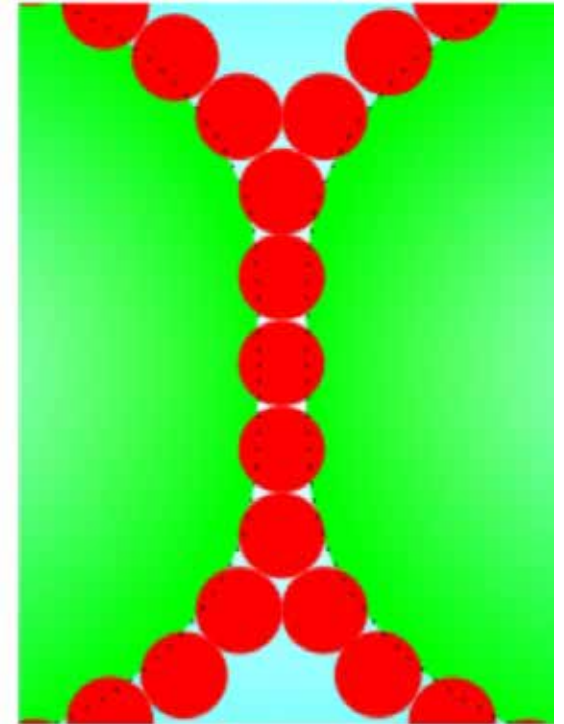
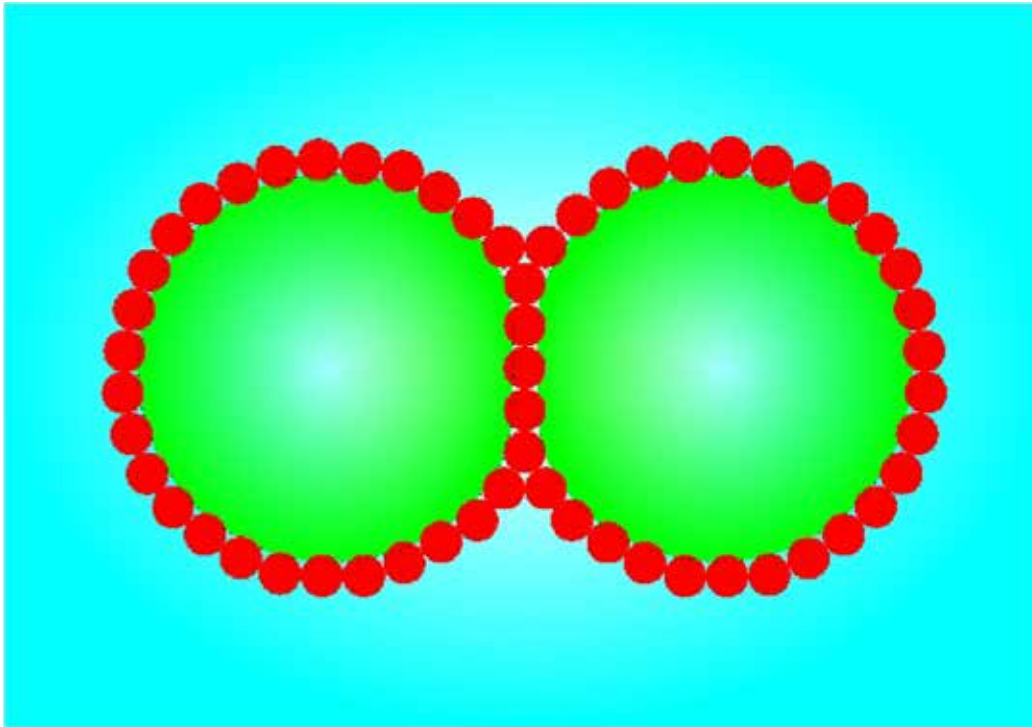
2. Making & breaking bridges



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# Colloidal particles are mesoscopic



This occurs when:

1. More interface is created than can be protected
2. The shear zone is small compared to the total volume
3. Particle protrude into the continuous phase

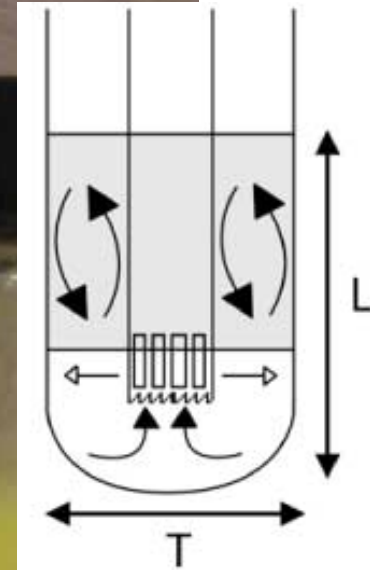
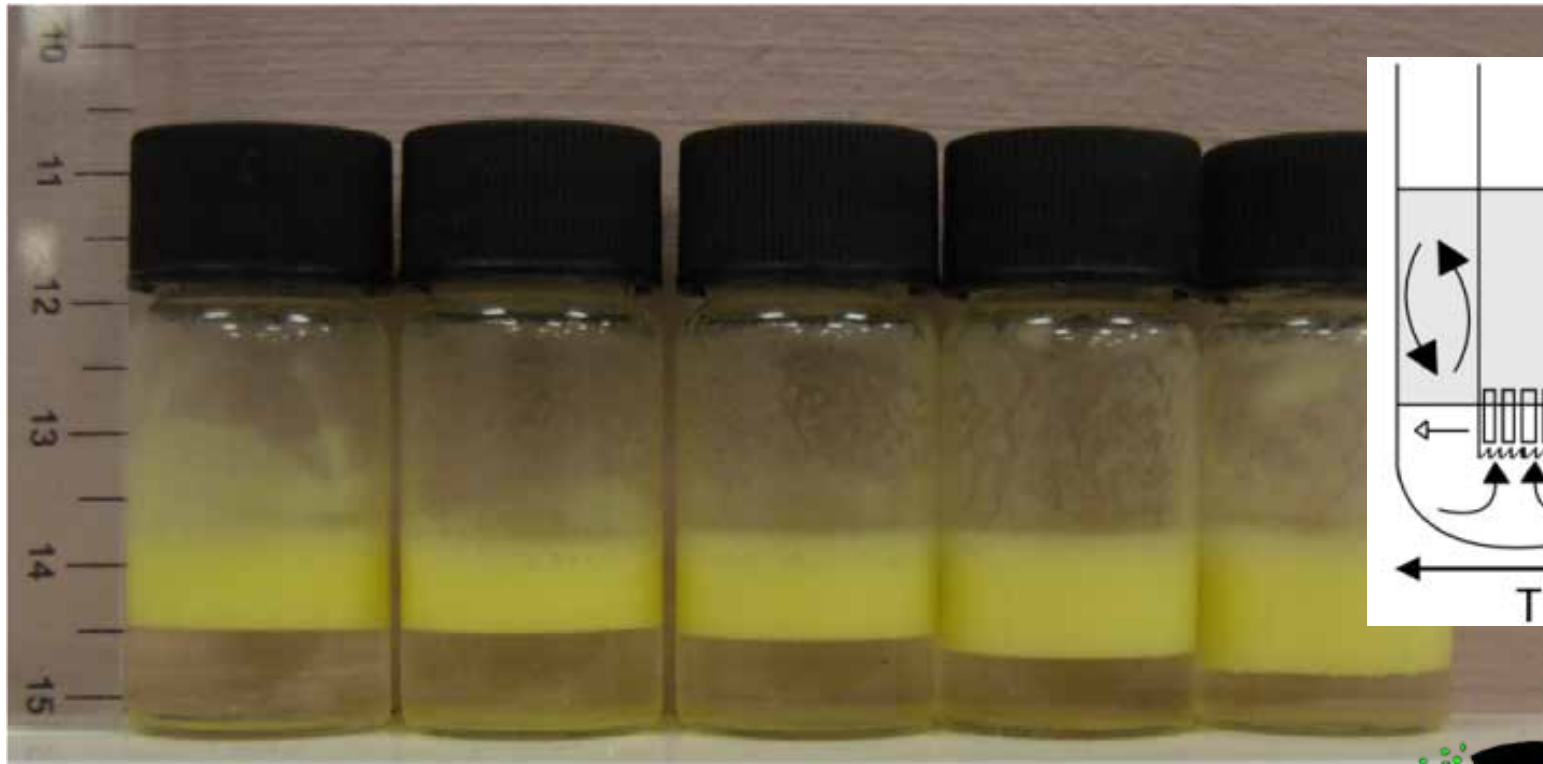


# Methods and materials

- Oil-in-water emulsions stabilised by Stober silica particles radius  $\approx 430\text{nm}$ .
- Oil phase is a mixture of dodecane and isopropyl myristate.
- Disperse particles in water with an ultrasonic probe - typically 150x 1s pulses.
- Vortex mix samples **until clear**.



# Shear rate



vortex  
only

5krpm

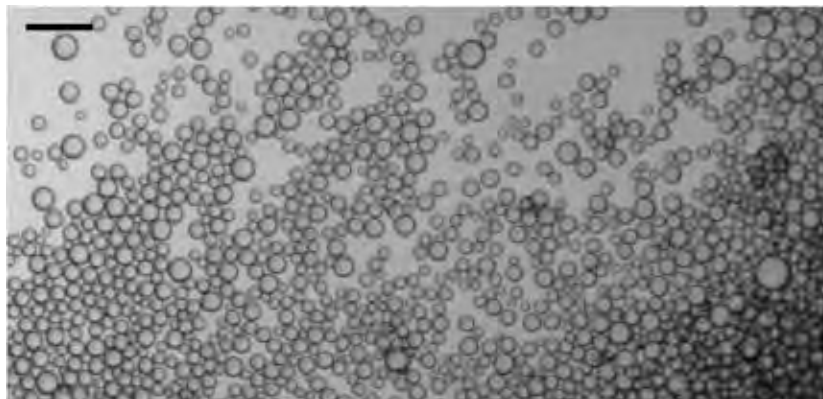
10krpm

15krpm

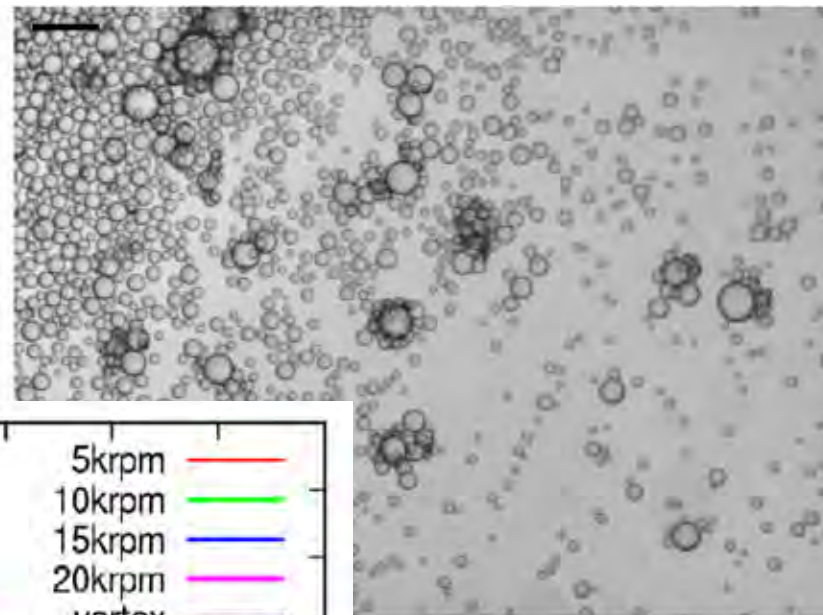
20krpm

$34,000 \text{ s}^{-1}$

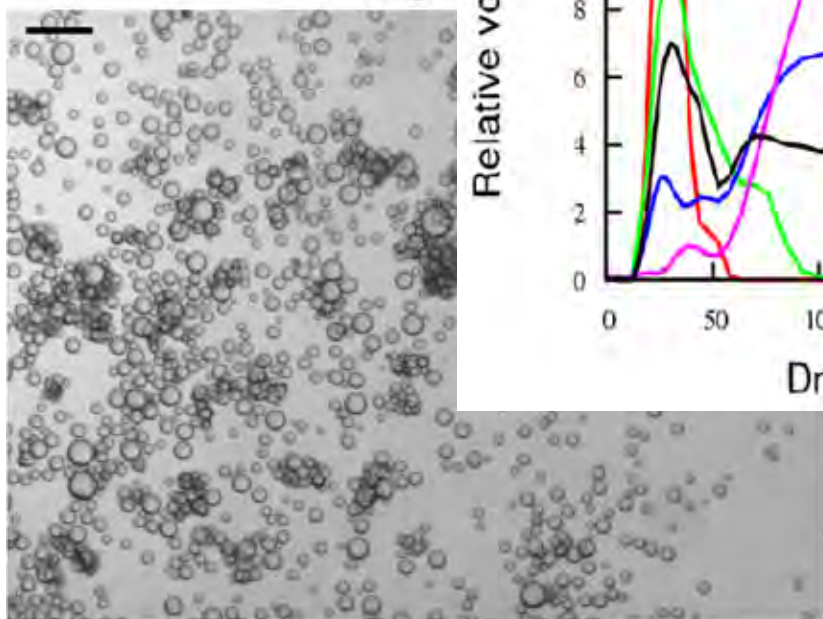




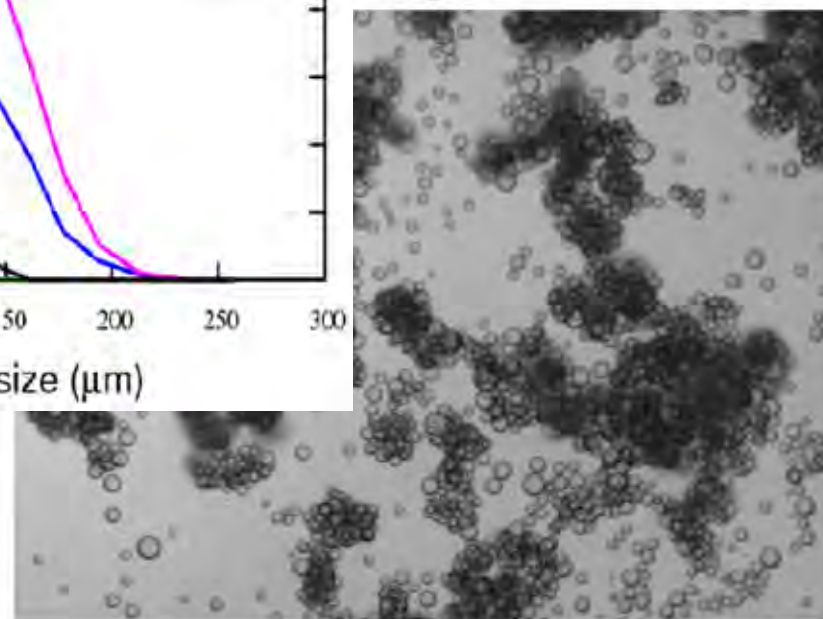
5krpm



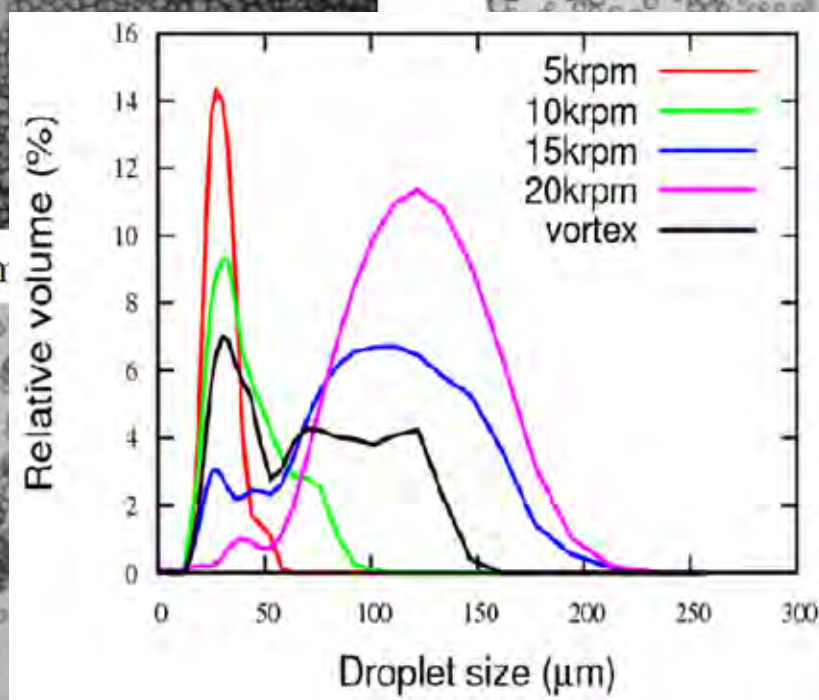
10krpm.



15krpm.



20krpm.





SEM shows that particles are indeed bridging droplets.

Also interesting are the hollows left on the droplet surfaces following fracture.

French et al JCIS 441, 30 (2015)

# Reshearing



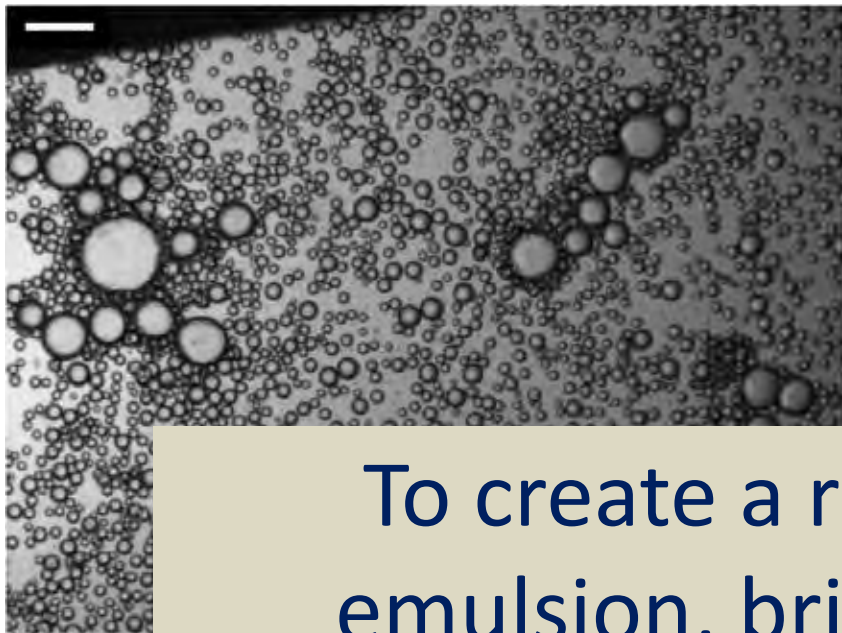
5krpm  
then  
20krpm

10krpm

15krpm

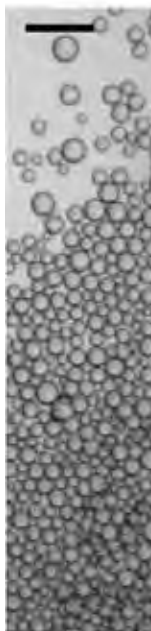
20krpm  
then  
5krpm

vortex  
only



To create a robust Pickering emulsion, bridging should be avoided

You need to use enough particles!

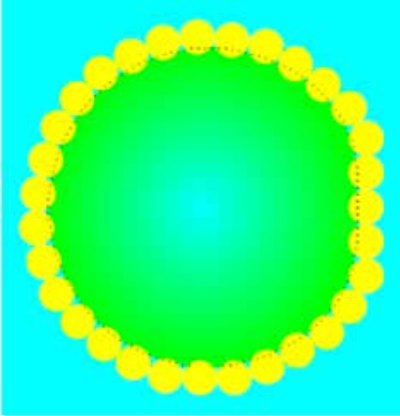


5krpm only.

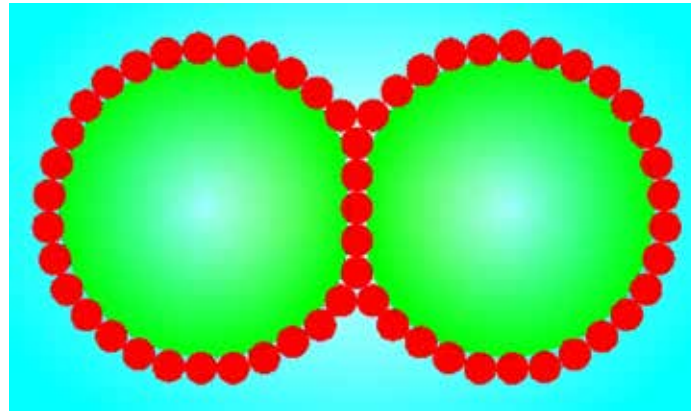
20krpm only.

# Outline

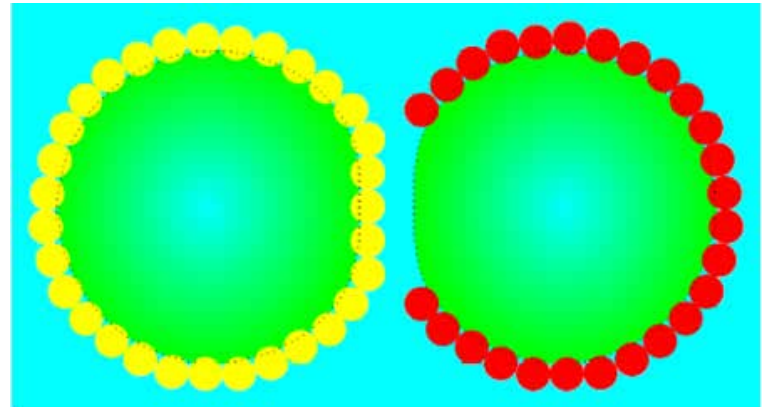
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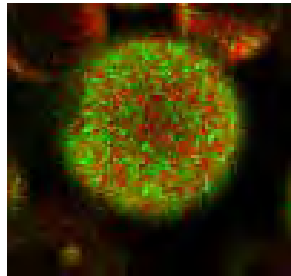
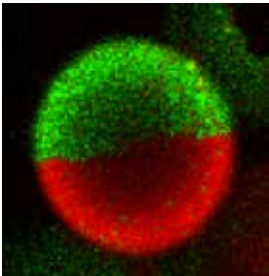


3. The secret life of  
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# Materials and Methods

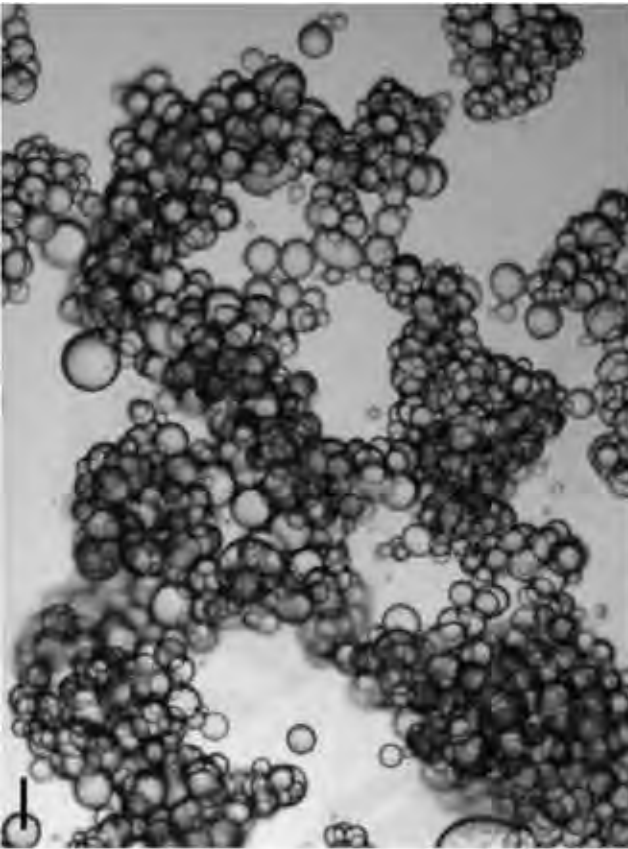
- Initially make two separate emulsions – one red, one green.
- Following emulsification, the red and green samples are mixed together as gently as possible.
- Place on a roller bank:



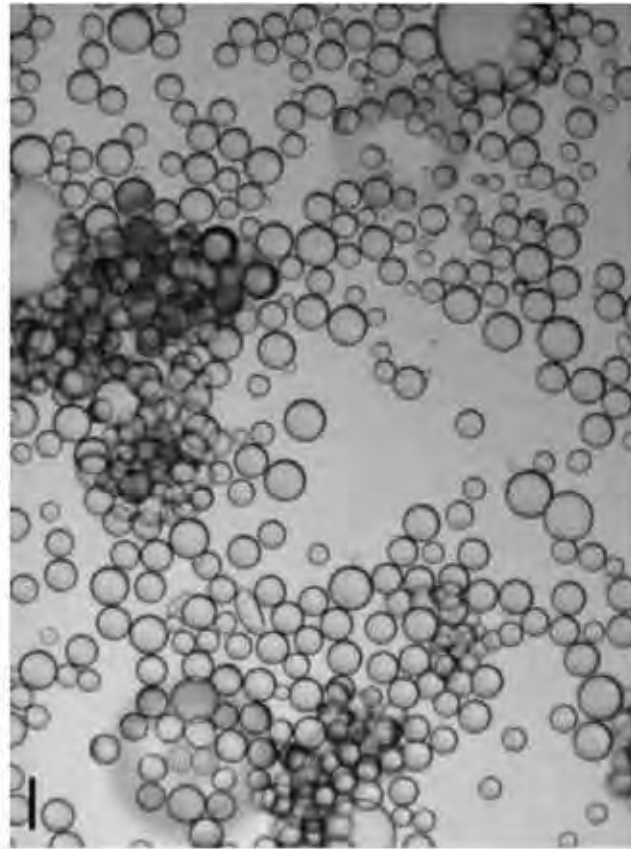
Shear rate  
 $3 \text{ s}^{-1}$



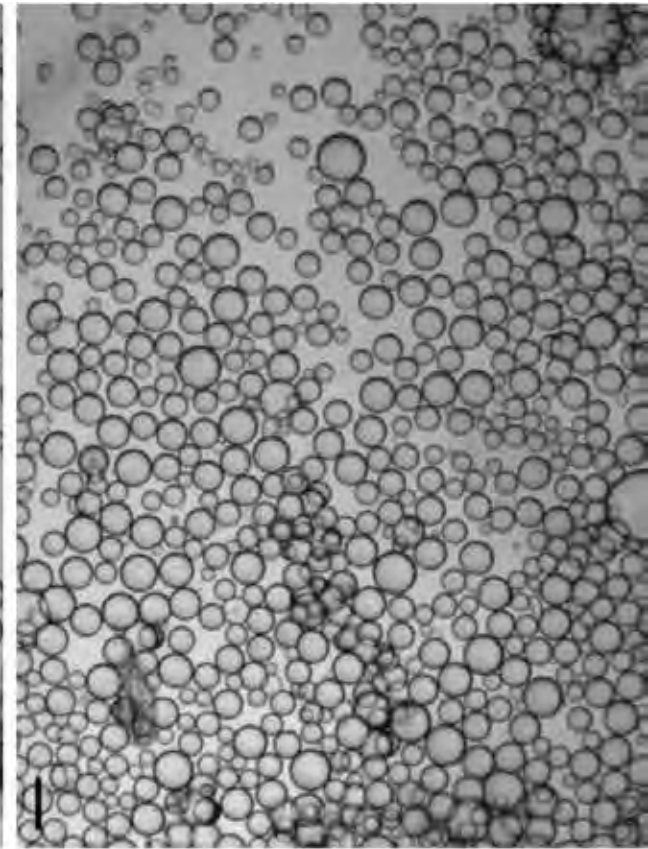
# Bright field microscopy



1 h



185 h



860 h

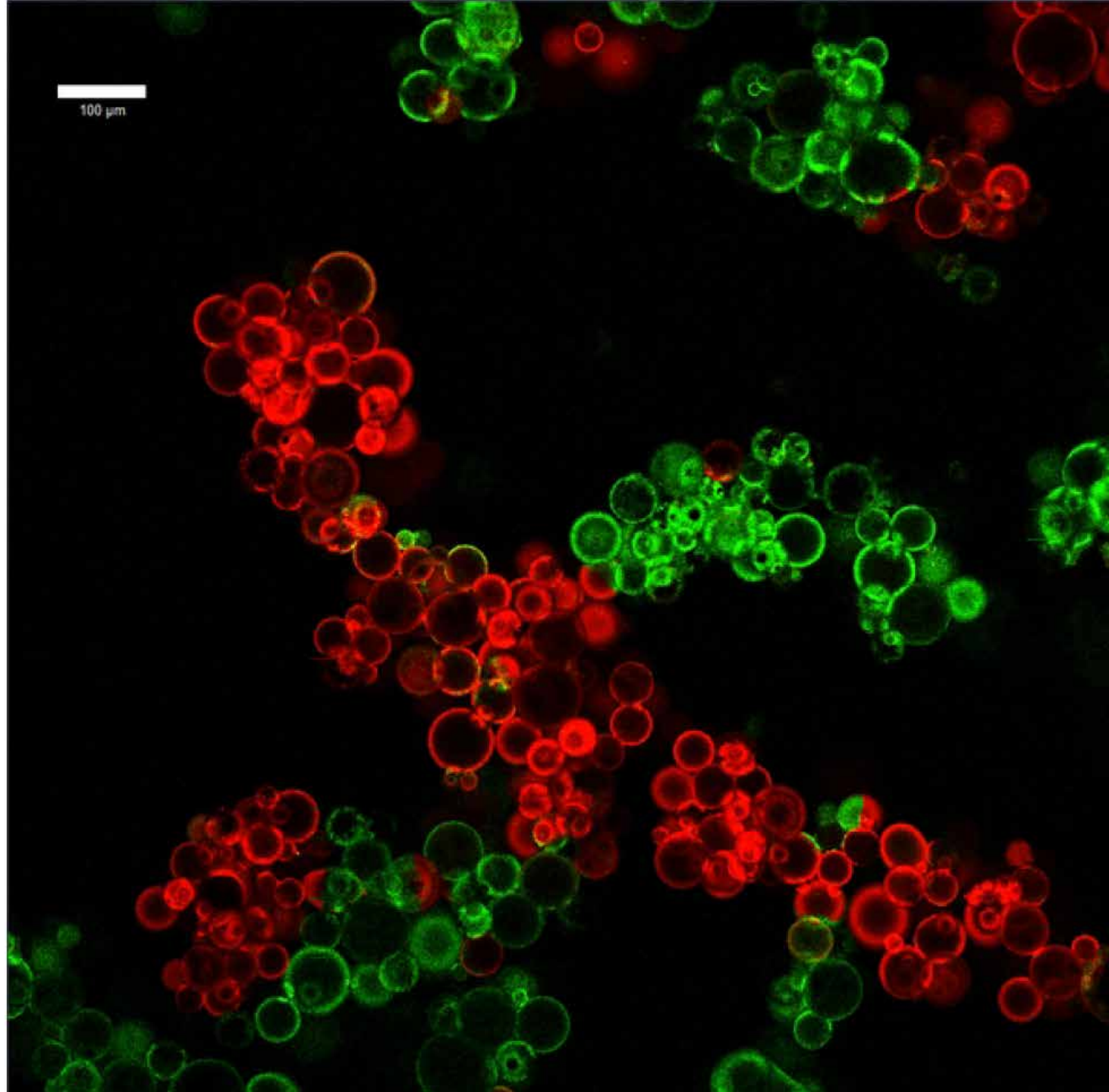
As the sample is rolled, the emulsion deaggregates. Scale bars are 100  $\mu\text{m}$ .

100  $\mu\text{m}$



**1 hour  
rolling**

Colours  
segregated

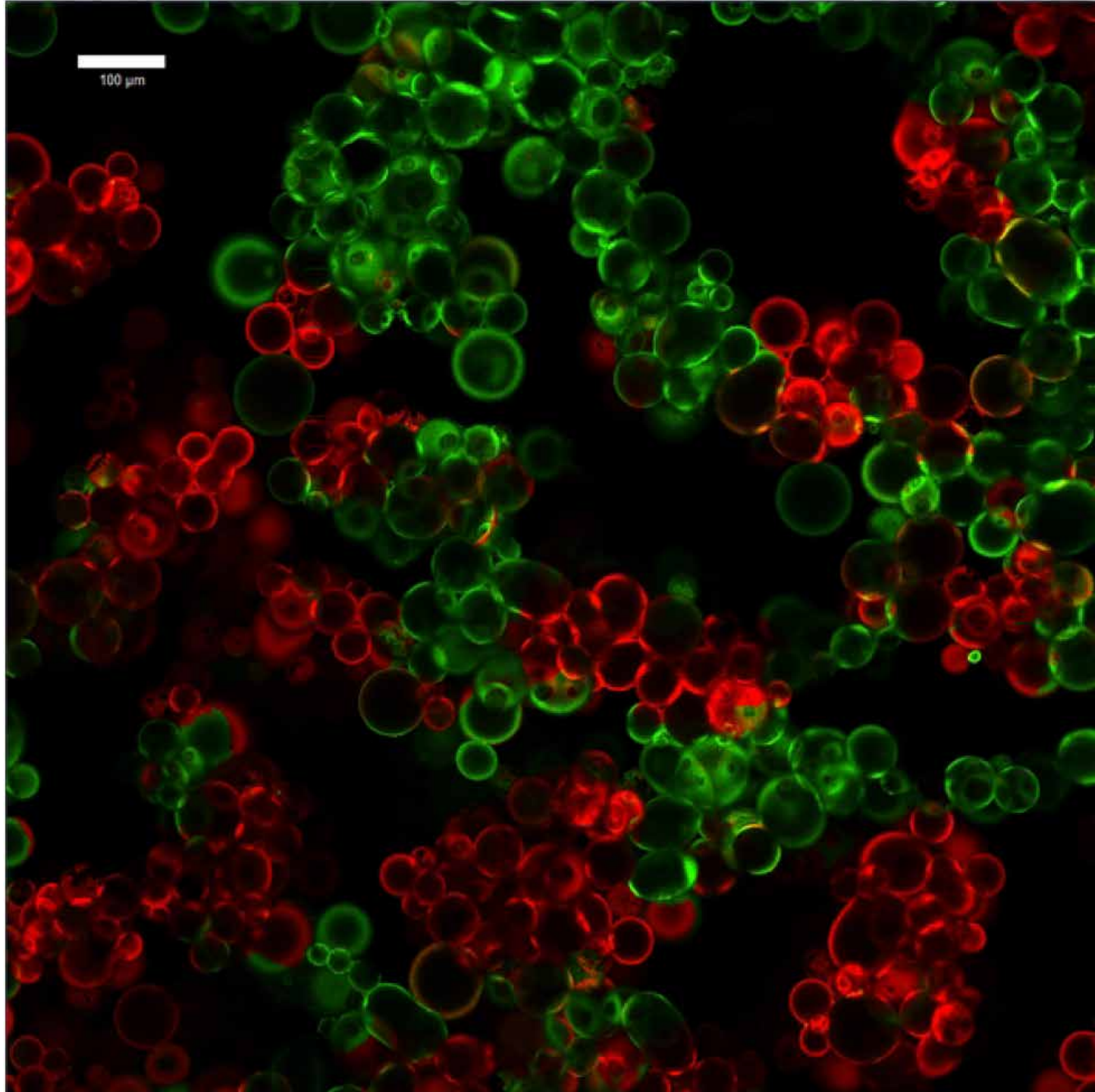


100  $\mu\text{m}$

6 hours  
rolling

More  
interspersed

Some multi-  
coloured drops



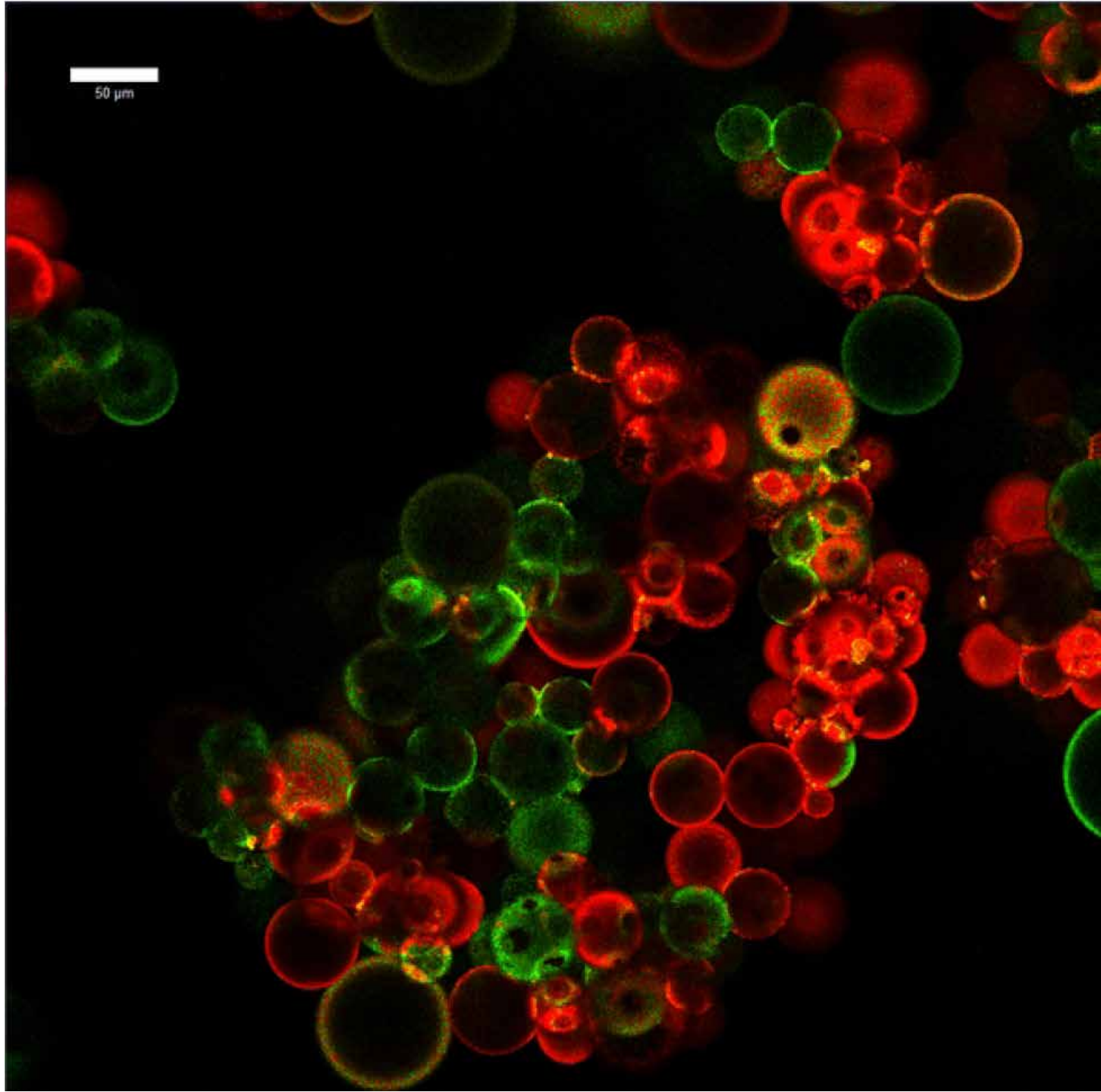
50  $\mu\text{m}$

50  $\mu\text{m}$

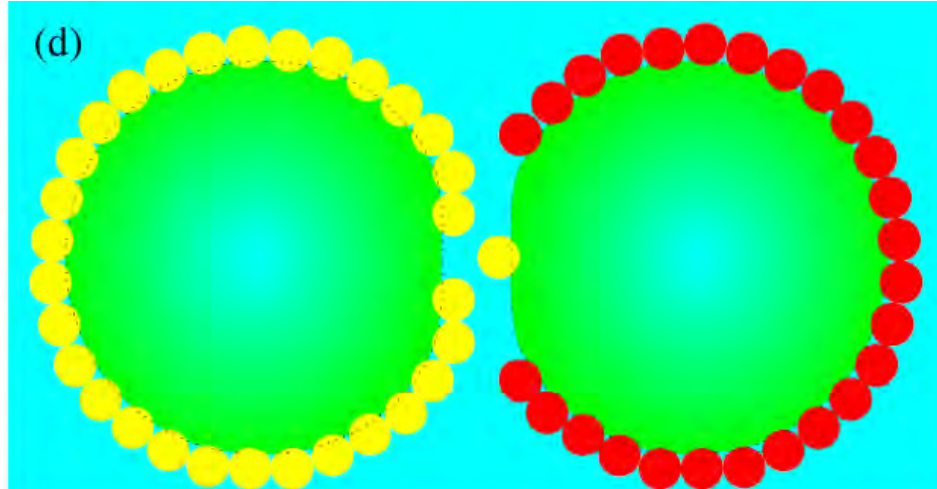
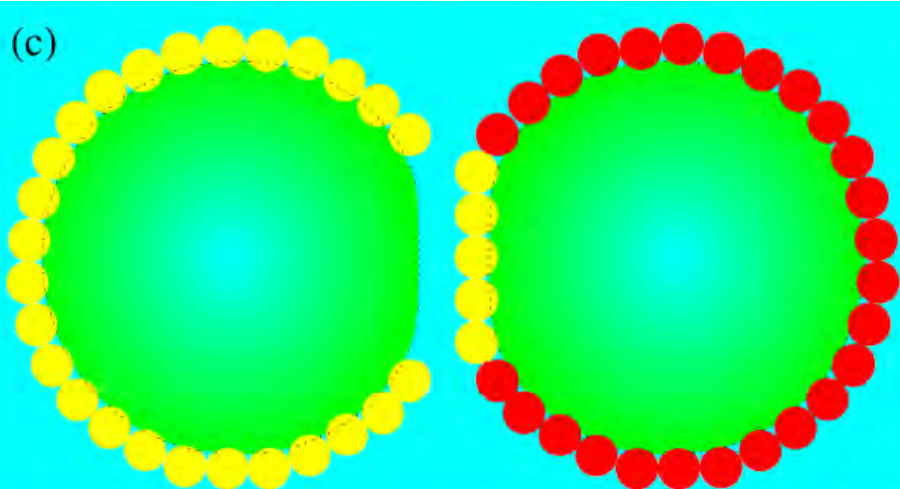
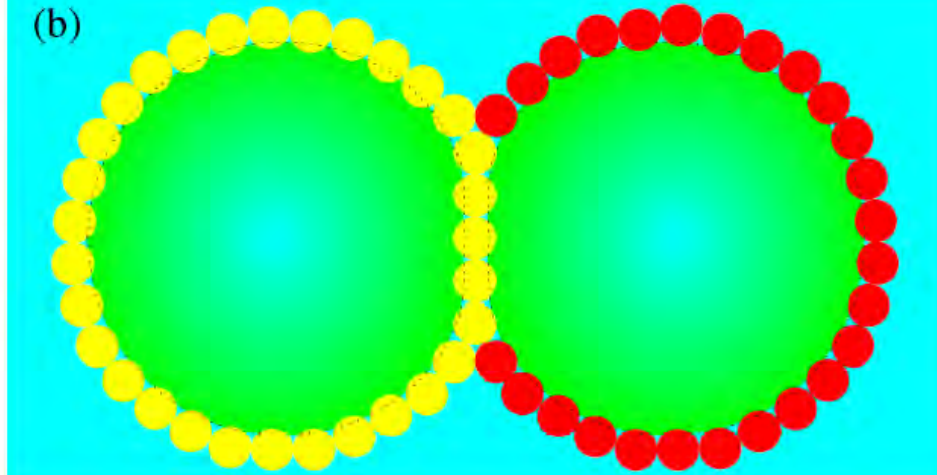
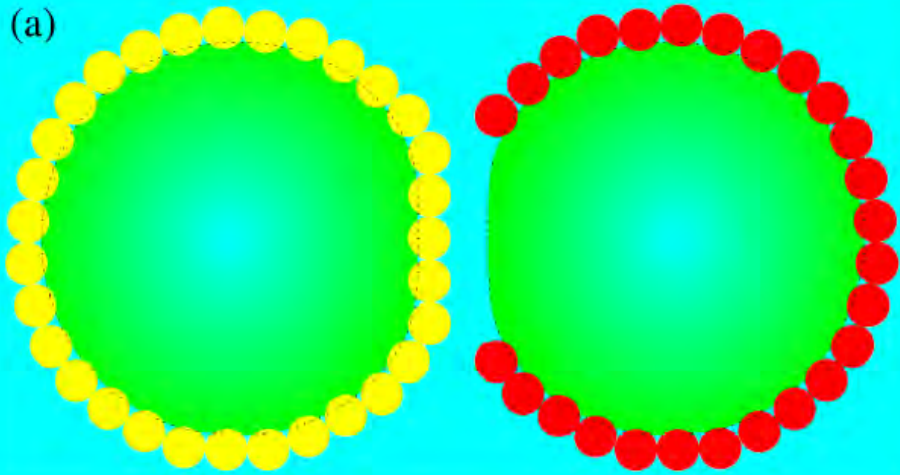
**21 hours  
rolling**

More thoroughly  
interspersed

Some well-mixed  
drops



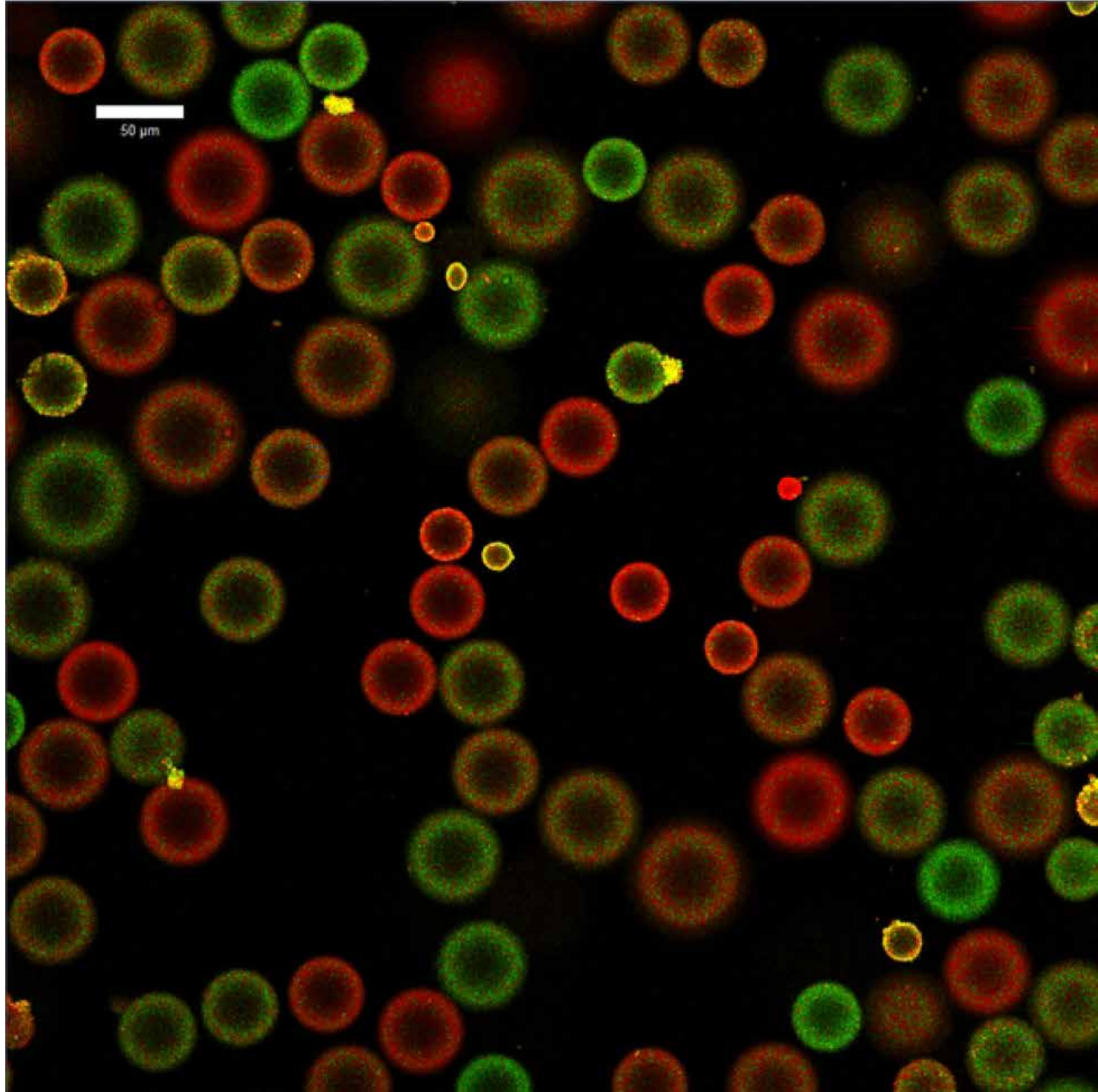
# Mechanism



50  $\mu\text{m}$

1191  
hours  
rolling

All multi-  
coloured drops



# Acknowledgements

- David French
- Phil Taylor (Syngenta, UK)
- Jeff Fowler (Syngenta, USA)
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- Joe Tavecchi
- Kathryn White

